NQR Study of the Phase Transition in 2SbCl3.o-Xylene

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Temperature dependence of ¹²¹Sb NQR frequencies for ^{2SbCl}₃·o-xylene showed that a phase transition takes place at about 139 K, whereas the DTA experiment showed no thermal anomaly around the transition temperature.

The formation of the 2:1 complexes between SbCl₃ and o-, m-, and p-xylene was reported as the results of phase diagram studies. We studied the bond characters between SbCl₃ and xylene in 2SbCl₃·p-xylene² and 2SbCl₃·m-xylene³ by means of nuclear quadrupole resonance(NQR). For 2SbCl₃·o-xylene, we obtained the different NQR results from the reported ones, i.e., six NQR lines were observed at 77 K in contrast to one resonance line reported before and from the temperature dependence of NQR frequencies a phase transition was found as stated below.

2SbCl₃·o-xylene was obtained by mixing SbCl₃ with o-xylene in the 2:1 molar ratio. The melting point of this complex was 35 °C(33.5 °C according to Ref. 1). NQR spectra were displayed on a pen-recorder by using a superregenerative oscillator and a lock-in amplifier and were also detected by a Matec pulsed NQR spectrometer(Model 5100 + 525 + 625). The temperature dependence of NQR frequencies was measured for the sample immersed into a bath of petroleum ether cooled by liquid nitrogen. The differential thermal analysis(DTA) was performed between 98 and 210 K using a homemade apparatus.

NQR parameters for $2\text{SbCl}_3 \cdot \text{o}-\text{xylene}$ are listed in Table 1. The quadrupole coupling constant $e^2\text{Qq/h}$ and the asymmetry parameter $_{1}$ were derived from the ratio of NQR frequencies at about 280 K. Since resonance lines except for ν_{1} (ν_{1} stands for the resonance line due to the transition $\text{m}=\pm 1/2 - \text{m}=\pm 3/2$) for ^{121}Sb and ^{123}Sb were too weak at 77 K to determine the frequency accurately, we could not assign a pair of ^{121}Sb and ^{123}Sb NQR lines to each equivalent atom in the crystal. The one ^{121}Sb NQR line was observed for $2\text{SbCl}_{3} \cdot \text{p}-\text{xylene},^2$) which indicates that the asymmetric unit in the crystal corresponds to one SbCl_{3} molecule and a half p-xylene molecule with the inversion center. 5 Although no x-ray analysis is available for $2\text{SbCl}_{3} \cdot \text{o}-\text{xylene}$, the unit which consists of two SbCl_{3} molecules and one o-xylene molecule without the inversion center is basically asymmetric one and then two Sb atoms in this unit have a different coordination environment and, therefore, two units like this exist in the crystal at room temperature considering the number of the NQR lines. The temperature dependence of $^{121}\text{Sb}(\nu_{1})$ NQR frequencies is shown in

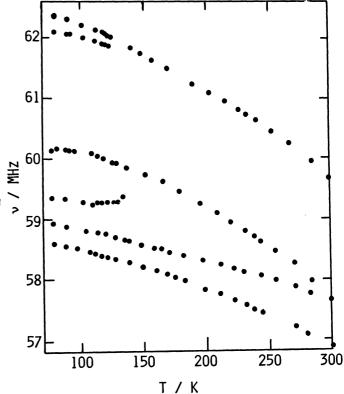
Fig. 1. Six NQR lines were observed at 77 K with the intensity ratios of 2:1:1:2:3:3 in order of decreasing frequency. On raising the temperature, the number of the six resonance lines reduced to four at about 139 K with the intensity ratios of 1:1:1:1. This means that the structural phase transition takes place at about 139 K and the symmetry of the crystal structure becomes the higher one above the transition temperature. No thermal anomaly appeared on the DTA curves around the transition temperature in the heating process. We tried to observe the free induction decay(FID) of $^{121}\mathrm{Sb}$ NQR below the transition temperature by using a pulsed NQR spectrometer. On raising the temperature the FID's of four higher-frequency lines were able to be observed up to ca. 90 K and for the rest up to ca. 115 K. The disappearance of the FID signal may be caused through the nonsecular part of the time constant T2 by the shortening of the spin-lattice relaxation time T_1 . We are planning to measure the temperature dependence of T₁'s in order to clarify this point. We consider that the only one line, which was relatively strong, of the six NQR lines observed at 77 K was reported as the early NQR results. 4)

Fig. 1. Temperature dependence of $^{121}\text{Sb}(\nu_1)$ NQR frequencies.

Table 1. NQR Parameters for 2ShCl₃·o-Xylene

Temperature		Frequency/MHz	
77 K	121 _{Sb(v₁)}	62.27, 62.10, 60.14 59.38, 58.92, 58.59	
	123 _{Sb(v1})	40.13, 39.34, 37.96 36.12, 35.81 ^{b)}	i
ca. 280 K	121 _{Sb(v1}) 123 _{Sb(v1})	60.12 e ² Qq/h/MHz 38.35 391.98	າ 0.146
	121 _{Sb(v1}) 123 _{Sb(v1})	58.13 36.93 379.52	0.140
	121 _{Sb(v1)} 123 _{Sb(v1)}	57.83 35.17 385.25	0.027
	121 _{Sb(v1)} 123 _{Sb(v1)}	57.12 35.51 378.90	0.067

- a) The experimental errors are within ±0.02 MHz.
- b) Doublet.



References

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